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# Synthesis, Radiolabeling and Preliminary Biological Evaluation of Radiolabeled 5-Methyl-6-nitroquipazine, a Potential Radioligand for the Serotonin Transporter

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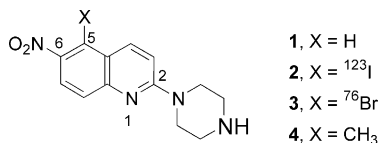
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**Abstract**—5-Methyl-6-nitroquipazine, a novel analogue of the potent and selective serotonin transporter inhibitor 6-nitroquipazine was synthesized and radiolabeled with tritium and the positron emitter carbon-11. [<sup>3</sup>H]5-methyl-6-nitroquipazine was found to have a  $K_d = 51 \pm 7$  pM. The high affinity and the facile labeling of [<sup>11</sup>C]5-methyl-6-nitroquipazine makes it a promising radioligand for visualization of the serotonin transporter with positron emission tomography.

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Neuronal uptake of serotonin (5-HT) occurs via a serotonin transporter protein (5-HTT) which plays a major role in the regulation of synaptic 5-HT levels. There are several arguments in favor of alterations of the 5-HTT in various neurological and psychiatric disorders such as Parkinson's and Alzheimer's disease and depression. In addition, the 5-HTT is the target of several antidepressant drugs. PET examination of the 5-HTT would be of great value in order to better understand the pathophysiological mechanisms of neurodegenerative and mental illness, as well as to quantify 5-HTT-occupancy in relation to antidepressant drug treatment. Several structurally diverse compounds such as [<sup>11</sup>C]McN5652, [<sup>123</sup>I]ADAM, [<sup>11</sup>C]DASB and [<sup>11</sup>C]RTI-357 have been suggested as potential radioligands for visualization of the 5-HTT with PET and SPECT (single photon emission computed tomography).<sup>1–4</sup>

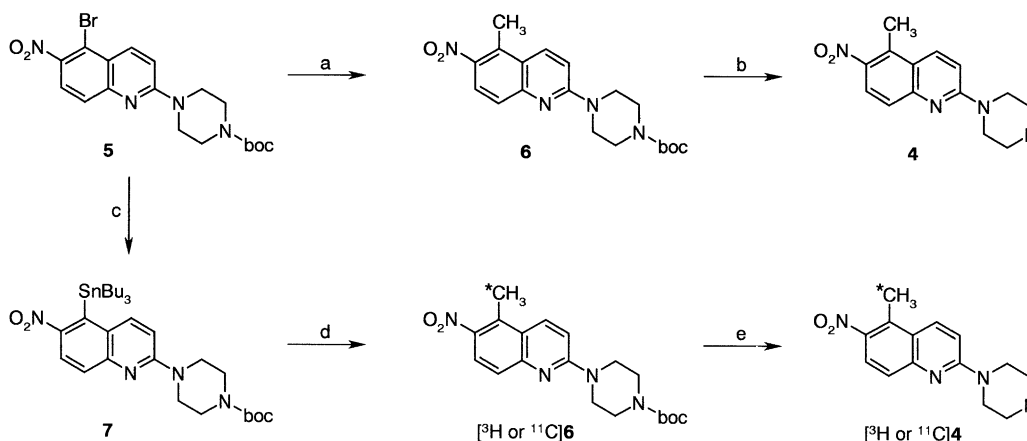


6-Nitroquipazine (**1**) is one of the most selective and potent 5-HTT inhibitors known.<sup>5–7</sup> Compound **1** has low affinity for non-serotonergic transporter systems and other pre- and postsynaptic receptor sites. The incorporation of the positron emitter carbon-11 ( $t_{1/2} = 20.4$  min) into the carbon framework of **1** is, however, not easily performed. The 5-position has been shown to be suitable for halogen substitution yielding halogenated analogues equipotent to **1**.<sup>8</sup> Iodine-123 labeled [<sup>123</sup>I]5-iodo-6-nitroquipazine (**2**) and bromine-76 labeled [<sup>76</sup>Br]5-bromo-6-nitroquipazine (**3**) have been reported as potential SPECT and PET radioligands, respectively.<sup>9,10</sup> Dehalogenation has been found to be a metabolic route for **2** in non-human primates. Consequently, SPECT images recorded with **2** are obscured by the presence of free iodine-123 in the primate brain.<sup>9</sup>

We hypothesized that a carbon-11 labeled and methylated analogue **4** would retain the affinity for the 5-HTT and that the absence of a halogen might be advantageous from a metabolic point of view. Furthermore, being less lipophilic, the non-halogenated [<sup>11</sup>C]**4** might be expected to give less non-specific binding in vivo than the halogenated counterparts **2** and **3**.

In this paper we report the synthesis and radiolabeling with tritium and carbon-11 of **4**. The binding affinity for

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**Scheme 1.** Reagents and conditions: (a)  $(\text{CH}_3)_4\text{Sn}$ ,  $\text{Pd}_2\text{dba}_3$ , DMF,  $70^\circ\text{C}$ , 24 h, 70%; (b) TFA,  $70^\circ\text{C}$ , 60 min, 95%; (c)  $n\text{-BuLi}$ ,  $\text{Bu}_3\text{SnCl}$ , THF,  $-100^\circ\text{C}$ , 2 h, 50%; (d) **7** (2.0 mg),  $[^3\text{H}/^{11}\text{C}]\text{MeI}$ ,  $\text{Pd}_2\text{dba}_3$  (0.9 mg),  $\text{P}(o\text{-CH}_3\text{C}_6\text{H}_4)_3$  (1.2 mg), DMF (350  $\mu\text{L}$ ),  $130^\circ\text{C}$ , 4 min, 80%; (e) TFA (150  $\mu\text{L}$ ),  $130^\circ\text{C}$ , 4 min, 95%. \* Position of label.

the 5-HTT was measured using rat frontal cortex homogenates and  $[^3\text{H}]\mathbf{4}$ .

### Chemistry

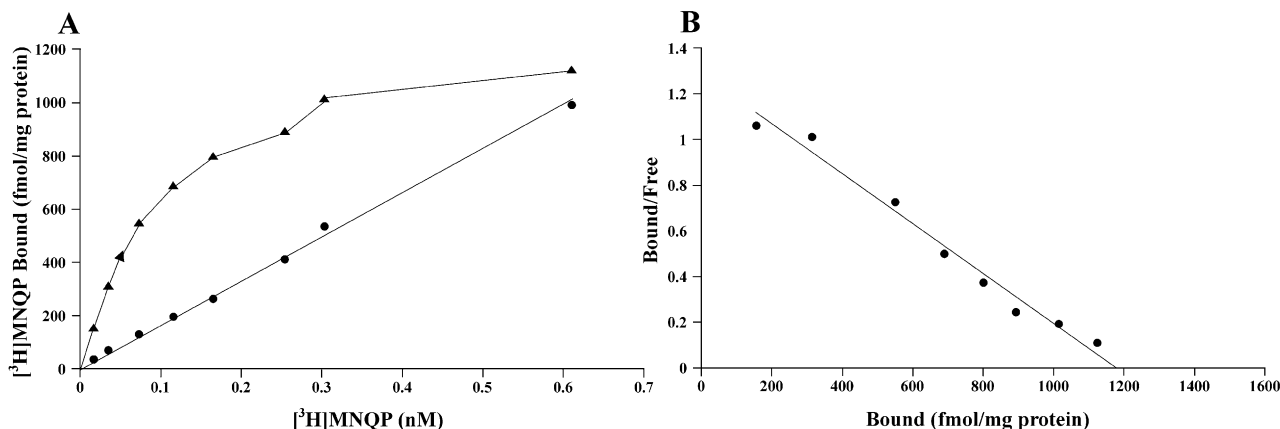
Compound **5** was prepared according to a procedure reported previously.<sup>11</sup> **6** was synthesized in a palladium promoted cross-coupling reaction between tetramethyltin and **5** (Scheme 1). Removal of the BOC-group with TFA yielded **4**.<sup>12</sup> Compound **7** was synthesized from **5** by lithiation and subsequent addition of tributyltinchloride.

### Radiochemistry

$[^3\text{H}]\mathbf{4}$  Was prepared from **7** in a two step procedure (Scheme 1). The first step was a palladium-promoted cross-coupling reaction between **7** and  $[^3\text{H}]\text{methyl iodide}$ .<sup>13</sup> The specific radioactivity of the  $[^3\text{H}]\text{methyl iodide}$  used was 3.15 GBq/ $\mu\text{mol}$ . Removal of the BOC-group was performed with TFA. Before purification by HPLC the reaction mixture was neutralized with aq/ $\text{NaOH}$ . Purification

by HPLC (Column C-18  $\mu\text{-Bondapak}$   $300\times 7.8$  mm; eluent methanol/water/triethylamine 63:37:0.2) yielded the final compound in 60% yield. The radioactive fraction containing  $[^3\text{H}]\mathbf{4}$  was collected and after evaporation of the mobile phase  $[^3\text{H}]\mathbf{4}$  was dissolved in ethanol and stored at  $-20^\circ\text{C}$  before use. Under these storage conditions no radiochemical degradation could be observed by HPLC analysis after 4 months.

The same protocol was applied for the radiolabeling with  $[^{11}\text{C}]\text{methyl iodide}$ . The  $[^{11}\text{C}]\text{methyl iodide}$  was produced from cyclotron produced  $[^{11}\text{C}]\text{carbon dioxide}$  using a GE MeI system.<sup>14</sup> The reaction mixture was purified by HPLC and the radioactive fraction containing  $[^{11}\text{C}]\mathbf{4}$  was collected and after evaporation of the mobile phase the residue was dissolved in sterile phosphate-buffered saline (pH = 7.4). The isolated yield of  $[^{11}\text{C}]\mathbf{4}$  was 60% (decay corrected, counted from the amount of  $[^{11}\text{C}]\text{methyl iodide}$ ). The total synthesis time was 40 min and the specific radioactivity was 25 GBq/ $\mu\text{mol}$  (range 19–30 GBq/ $\mu\text{mol}$ ,  $n=4$ ) at the end of the synthesis. The obtained radioligand was analyzed by HPLC showing a purity better than 99%.



**Figure 1.** Saturation experiments of  $[^3\text{H}]\text{MNQP}$  binding in the rat prefrontal cortex. Membranes were incubated in Tris-HCl buffer with various concentrations of  $[^3\text{H}]\text{MNQP}$  (30–1000 pM) for 90 min at  $22^\circ\text{C}$ . Non-specific binding was estimated in the presence of 1  $\mu\text{M}$  paroxetine. The results shown are from a typical experiment and the values are the mean of duplicate determinations. (A) The saturation binding isotherm shows specific binding (▲) and non-specific binding (●). (B) Scatchard plot analysis of binding of  $[^3\text{H}]\text{MNQP}$  resulted in a  $K_d$  of  $51\pm 7$  pM and a  $B_{\text{max}}$  of  $1530\pm 395$  fmol/mg protein.

## Results and Discussion

The binding assay was performed essentially as described by Bonnet et al.<sup>15</sup> Figure 1A shows a saturation curve of [<sup>3</sup>H]4 to rat frontal cortex homogenate. The linearity of the Scatchard plot in Figure 1B is indicative of the presence of a single class of binding sites with a  $K_d = 51 \pm 7$  pM and  $B_{max} = 1530 \pm 395$  fmol/mg protein ( $n=3$ , independent experiments each performed as duplicates). The dissociation constant of 4 is in the same range as for 1,<sup>6</sup> thus 4 and 1 are equipotent.

We recently reported a positron emission tomography examination of [<sup>11</sup>C]4 in cynomolgus monkey brain.<sup>16</sup> The regional distribution of radioactivity was in accordance with the known distribution of 5-HTT, and pre-treatment with the selective 5-HTT inhibitor citalopram indicated that [<sup>11</sup>C]4 binds specifically to 5-HTT in vivo in the primate brain. The high binding constant, specific in vivo binding, and the facile labeling makes [<sup>11</sup>C]4 a potential radioligand for visualization of the 5-HTT with PET. Furthermore, the low non-specific binding observed in vitro with [<sup>3</sup>H]4 renders this ligand a promising radioligand for in vitro studies of the 5-HTT.

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 12. Characterization of compound 4. Yellow solid, mp 122–125 °C, <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ: 1.75 (br s, 1H, NH), 2.79 (s, 3H, CH<sub>3</sub>), 3.02 (t,  $J=4.6$  Hz, 4H, CH<sub>2</sub>), 3.80 (t,  $J=4.6$  Hz, 4H, CH<sub>2</sub>), 7.06 (d,  $J=9.5$  Hz, 1H, H3), 7.54 (d,  $J=9.2$  Hz, 1H, H8), 8.00 (d,  $J=9.2$  Hz, 1H, H7), 8.22 (d,  $J=9.5$  Hz, 1H, H4).  
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